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### **Principles of Micro- and Nanofabrication for Electronic and Photonic Devices**

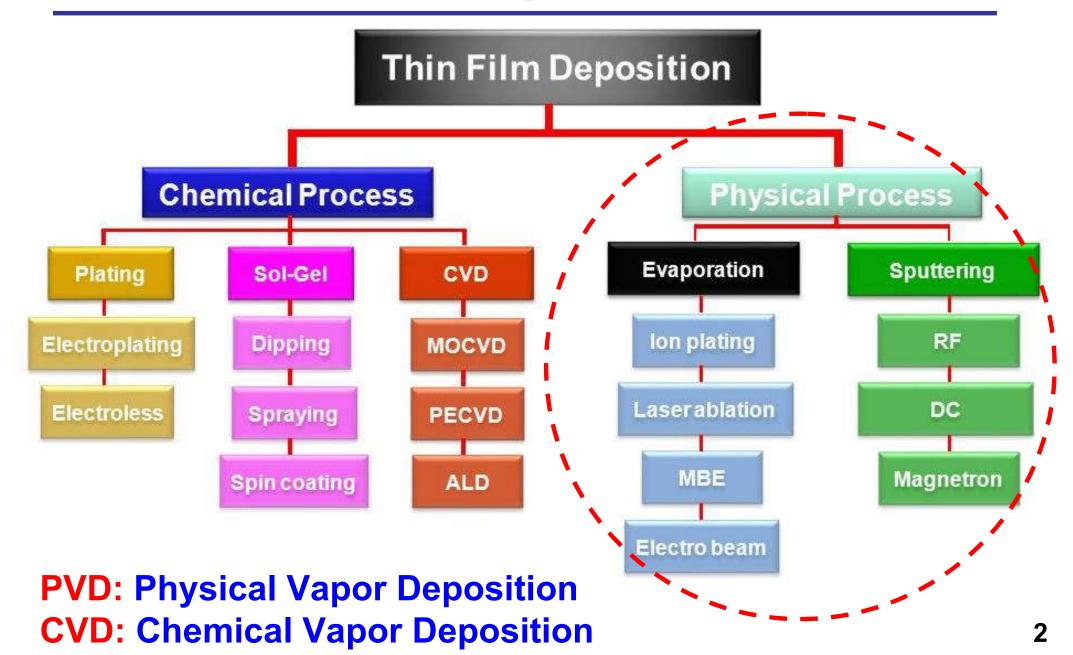
# Film Deposition Part V: PVD

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### **Film Deposition**

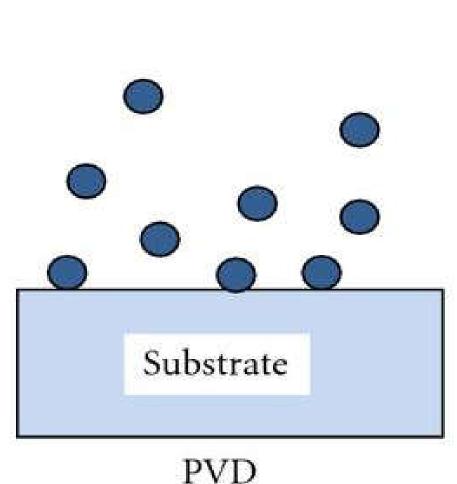


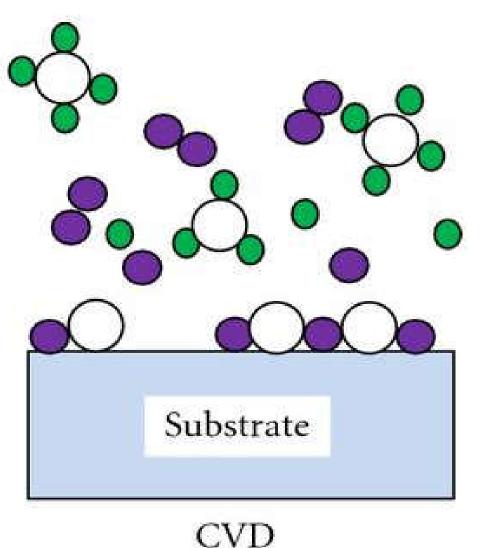
Xing Sheng, EE@Tsinghua

### **PVD vs. CVD**

#### **Physical process**

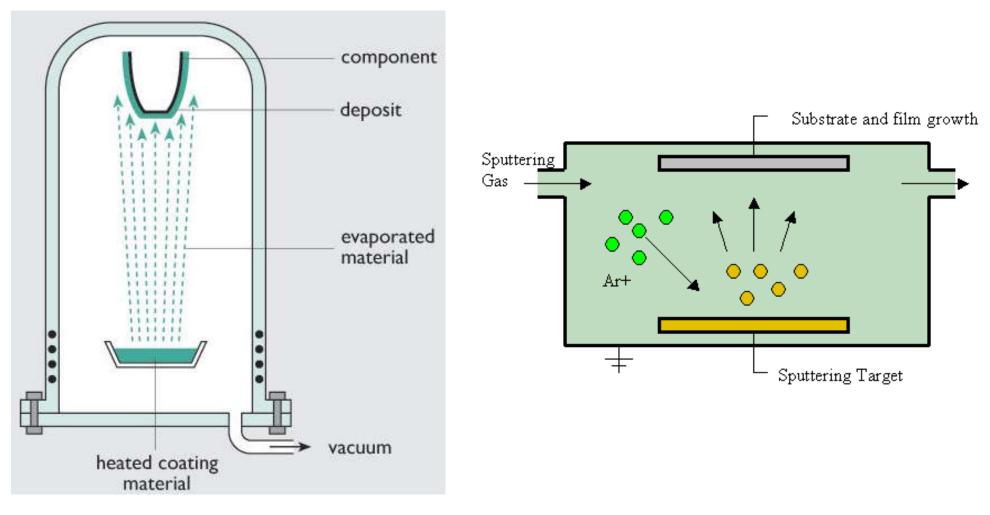
#### **Chemical reactions**





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### PVD



#### Evaporation (蒸发)

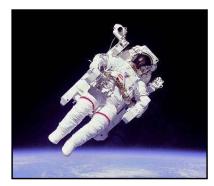
#### Sputter (溅射)

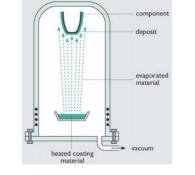
### **Vacuum Basics**

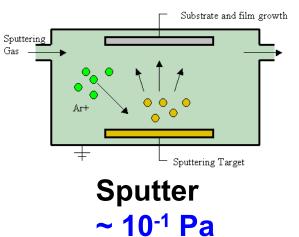
#### Units

- □ 1 Pa = 1 N/m<sup>2</sup>
- **1** atm = 760 torr = 760 mm Hg = 1.013\*10<sup>5</sup> Pa
- □ 1 bar = 10<sup>5</sup> Pa = 750 torr
- □ 1 torr = 133.3 Pa







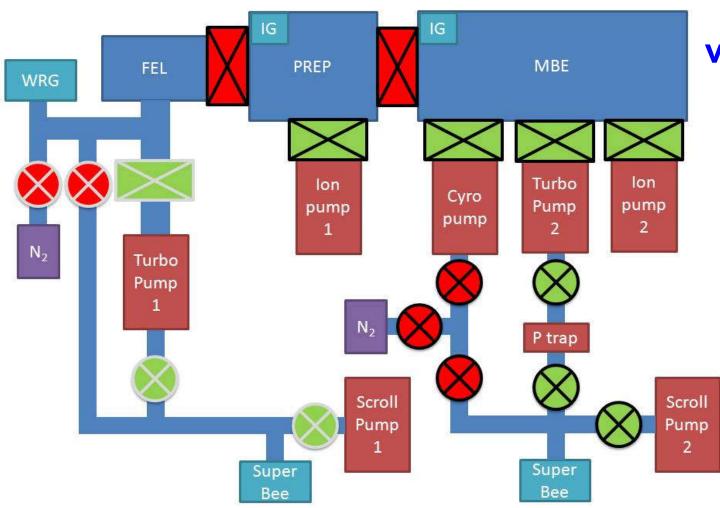


Pressure cooker ~ 1.5 atm

outer space < 10<sup>-10</sup> Pa

Evaporation < 10<sup>-7</sup> Pa

### **Vacuum Systems**

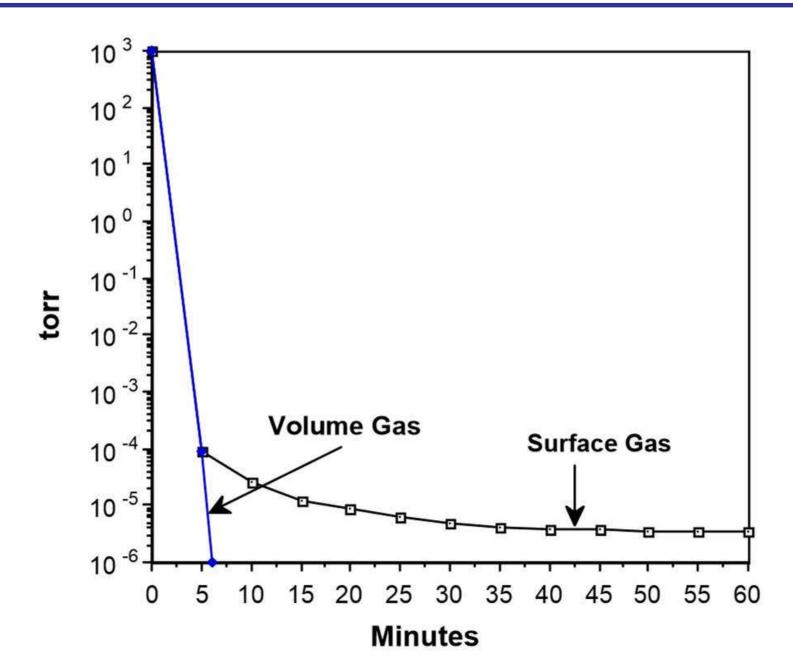


#### vacuum ~ 10<sup>-10</sup> Pa



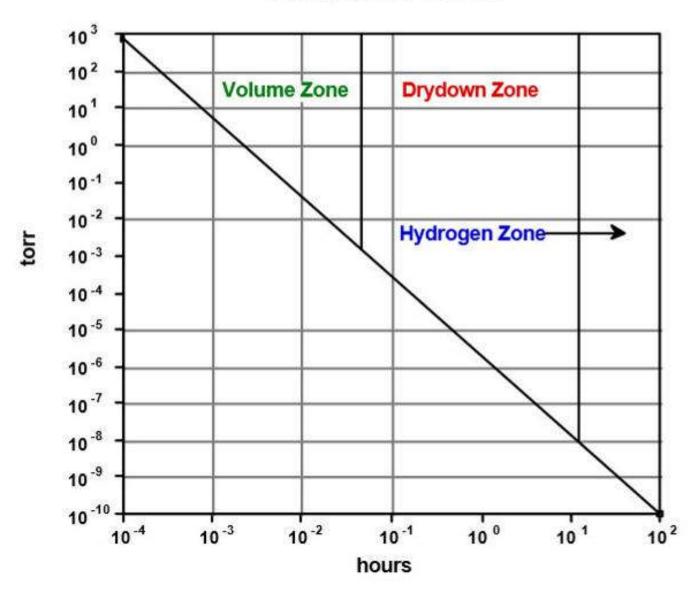
#### MBE: Molecular Beam Epitaxy 分子束外延

### Vacuum Pumpdown



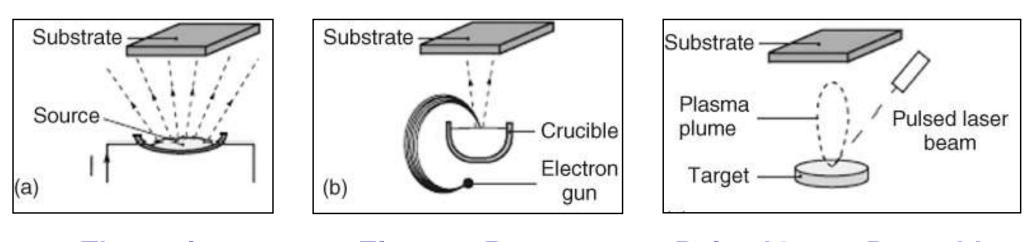
### Vacuum Pumpdown

**Pumpdown Zones** 



air
 water
 hydrogen

#### **Evaporation**



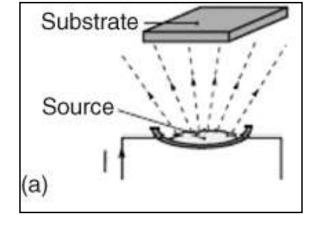
Thermal

Electron Beam (Ebeam) Pulsed Laser Deposition (PLD)

#### **Q:** Why do we need the vacuum?

### **Evaporation**

- Reduce the impurities (N<sub>2</sub>, O<sub>2</sub>, H<sub>2</sub>O, ...)
- Prevent oxidation
  e.g. Cu + O<sub>2</sub> -> CuO
- Ballistic transport
  - **n** molecular mean free path  $\lambda$

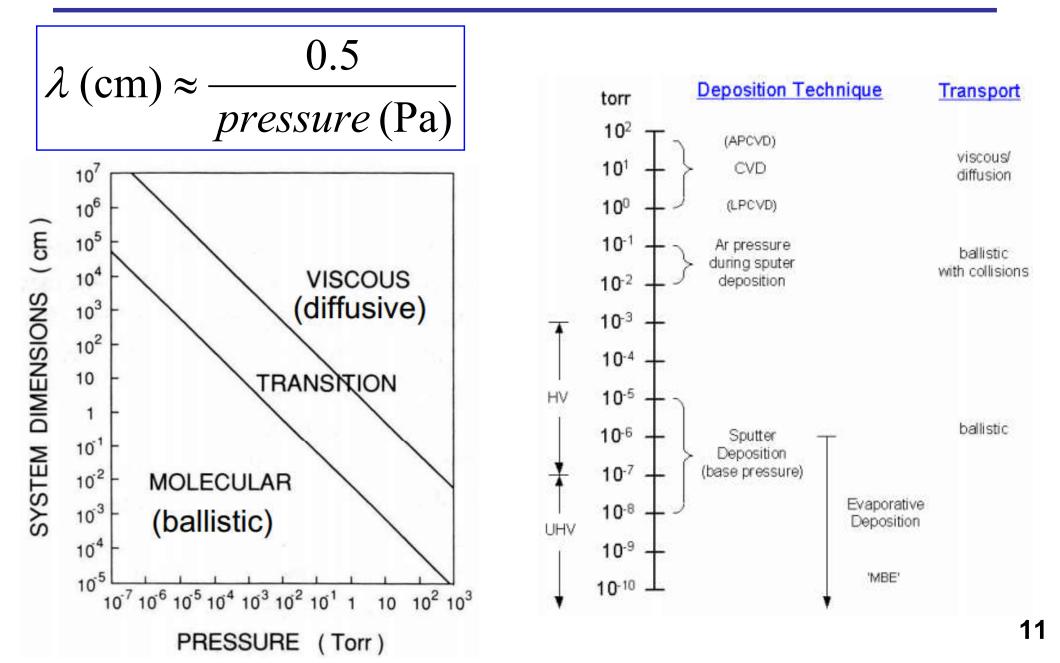


$$\lambda = \frac{kT}{\sqrt{2}\pi r^2 p}$$

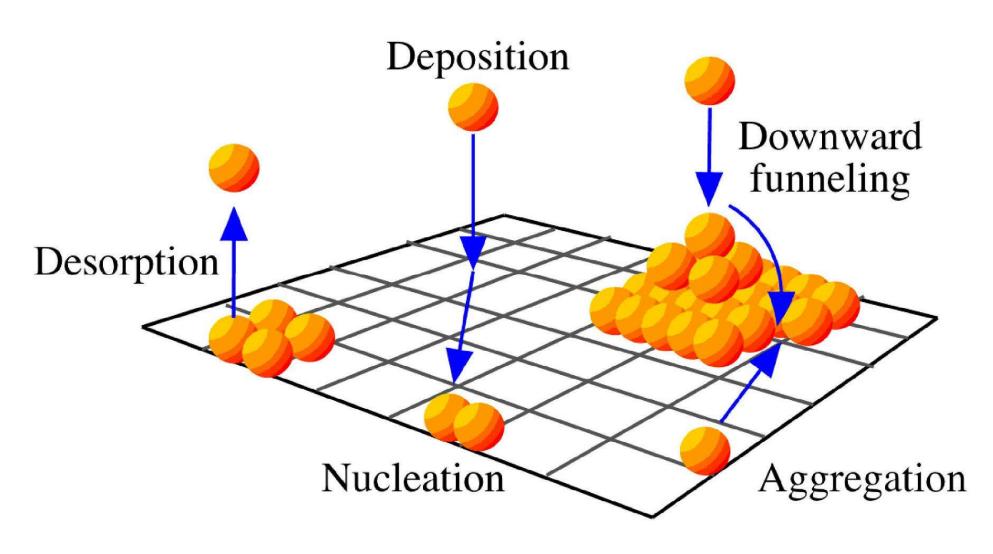
$$\lambda$$
 (cm)  $\approx \frac{0.5}{pressure (Pa)}$ 

#### **Q: Required pressure?**

### **Mass Transport**



### **Mass Transport**



absorption - movement - desorption

### **Evaporation Rate**

#### Langmuir-Knudsen Theory

$$R_{evap} = 5.83 \times 10^{-2} A_s \sqrt{\frac{m}{T}} P_e$$

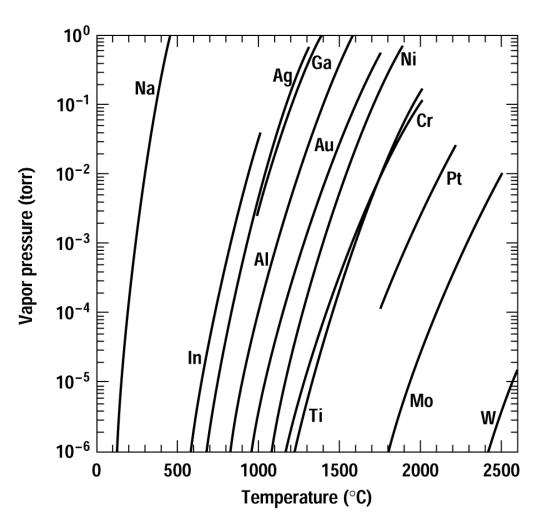
**R**<sub>evap</sub>: Evaporation rate (g/s)

 $A_s$ : area of sources (cm<sup>2</sup>)

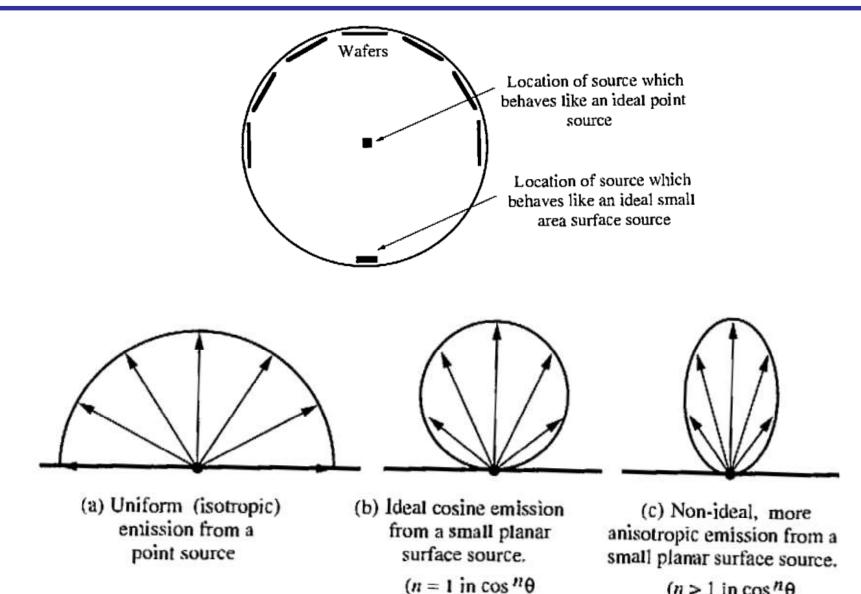
m: molecular weight (g/mol)

T: temperature (K)

**P**<sub>e</sub>: vapor pressure of sources (Torr) (*not* chamber pressure)



### **Evaporation Sources**

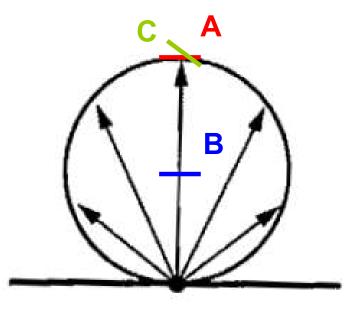


distribution)

 $(n > 1 \text{ in } \cos^n \theta$ distribution)

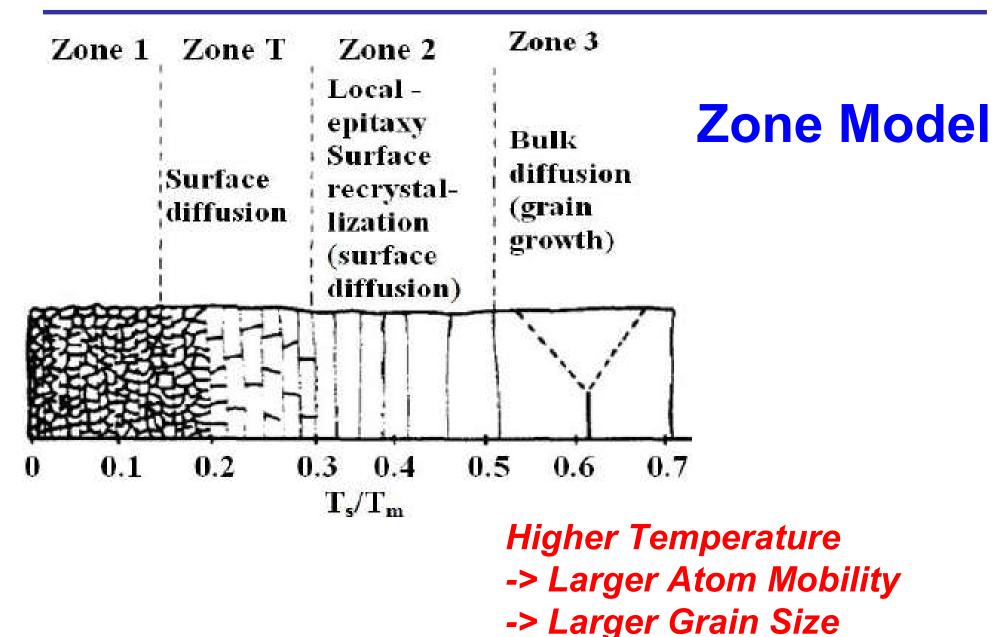
### **Deposition rate**

# Question: 1. R<sub>A</sub> : R<sub>B</sub> = ? 2. R<sub>A</sub> : R<sub>C</sub> = ?

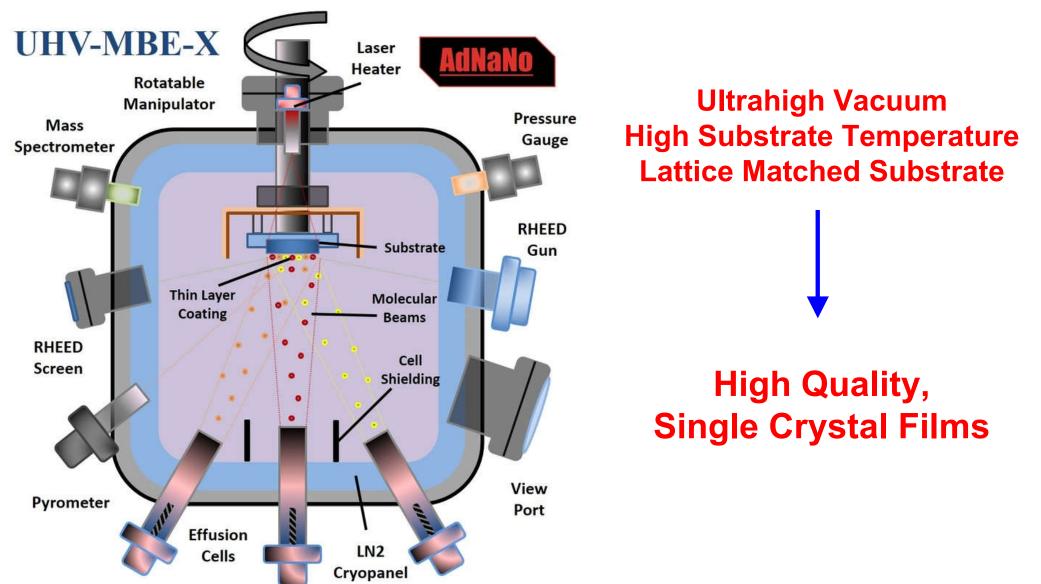


Ideal cosine emission from a small planar surface source.

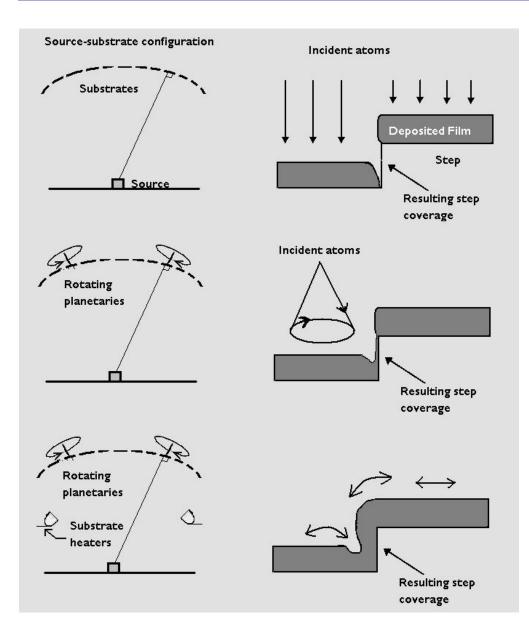
#### **Effects of Substrate Temperature**



### **MBE: Molecular Beam Epitaxy**



### **Step Coverage**



Substrate rotation and heating improve step coverage

### **Challenges of Evaporation**

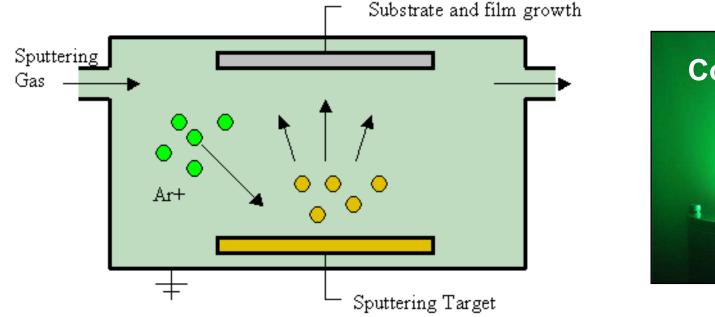
Materials with high melting points / low vapor pressure

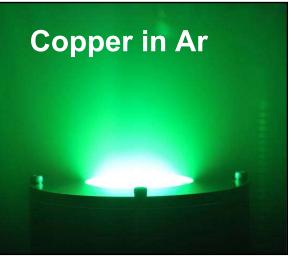
- $\square$  W, Mo, SiO<sub>2</sub>, ...
- Compounds and alloys (non-stoichiometry)
  FeCoB alloy
  TiO<sub>2</sub> -> TiO<sub>x</sub>
- Radiation damage generated by Ebeam
  electron beam and X-ray radiation
- Poor step coverage
  - via filling



#### Plasma (e.g. Ar) assisted transport

- high energy
- high deposition rate



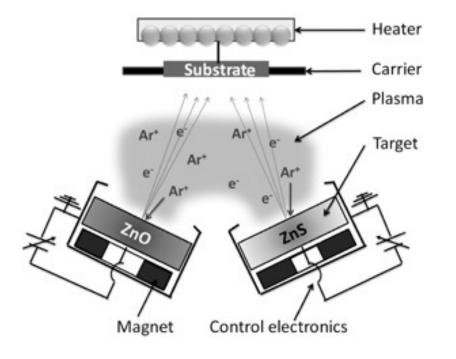




#### **Co-Sputter**

#### Deposit more than one material

**composition control** 





### Sputter: Pros. & Cons.

#### Advantages

- **Higher pressure than evaporation**
- Higher deposition rate
- **Better uniformity and step coverage**
- Better stoichiometry control
- Work for most materials

#### Disadvantages

- Plasma induced damages (etching)
- More impurities and defects
- Not good for single crystal epitaxy
- Mostly polycrystalline and amorphous films

### **Sputter**

#### Process Parameters

- **Type: DC, RF/AC, Magnetron, ...**
- Substrate temperature
- □ Gas type (Ar, O<sub>2</sub>, N<sub>2</sub>, ...)
- Chamber pressure
- Sputter power
- ••••

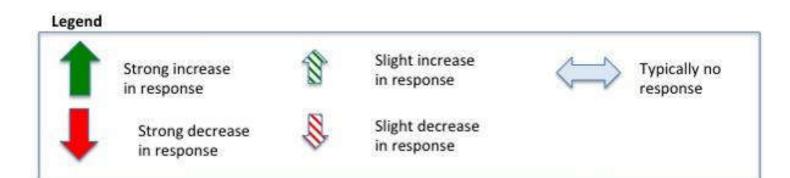
#### Control Parameters

- **Deposition rate**
- Crystallinity
- □ Film quality (defects, ...)

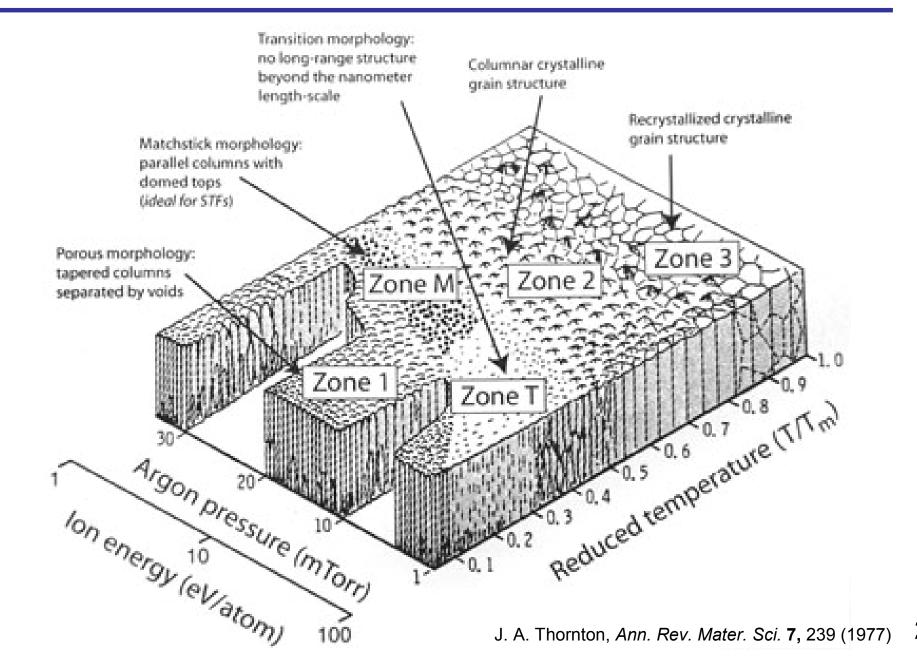
### **Sputter**

#### Sputtering Process Trend for typical metals and films

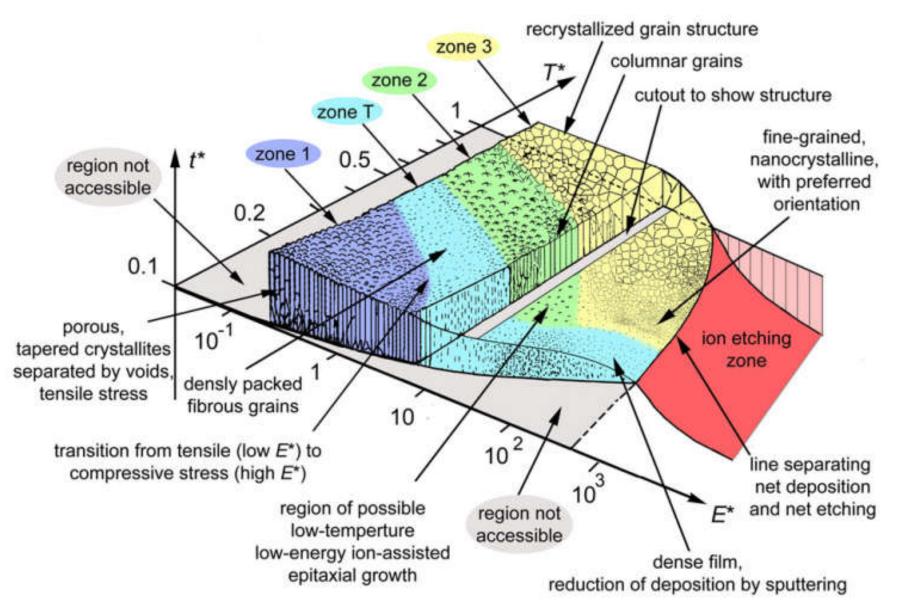
	Base Pressure	Sputtering Pressure	Power	Substrate RF Bias
Deposition Rate	$\langle \downarrow \rangle$	Below"3mT Above"8mT	1	8
Stress (+ tensile, - compressive)	$\langle \rangle$	1	Ļ	Ļ
Step Coverage/ Sidewall coverage	$\langle $	1	2 <sup>nd</sup> order effect depending on geometry	Can cause re-dep onto sidewalls thru collisions
Resistivity	1		2 <sup>nd</sup> order effect with substrate or target heating on some films	2 <sup>nd</sup> order effect with some films by changing density or stress



### **Thornton's Zone Model**



### **Refined Zone Model**



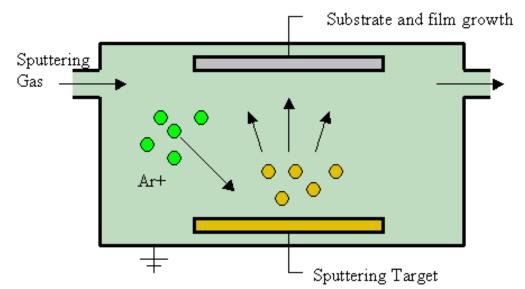
### **Evaporation vs. Sputter**

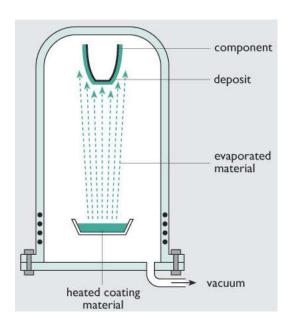
#### **Evaporation:**

- higher temperature
- radiation (Ebeam)
- lower pressure
- poor step coverage

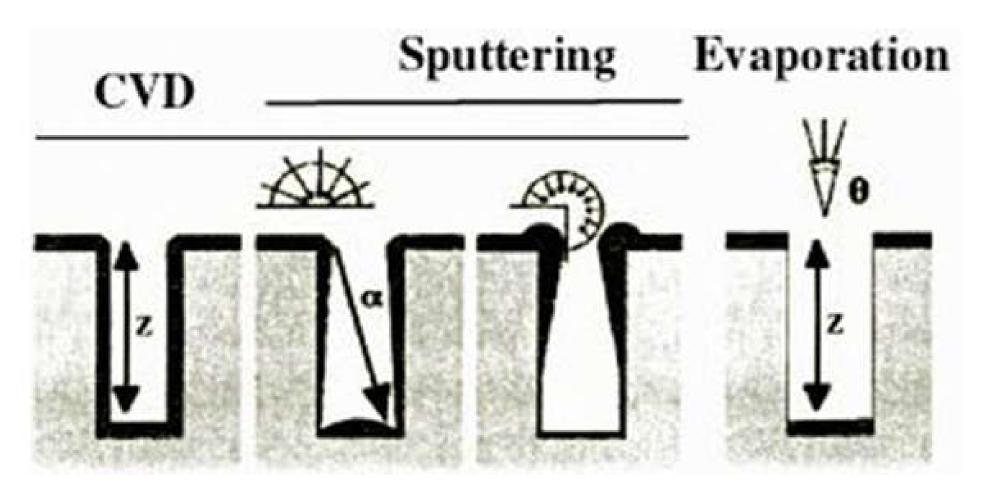
Sputter:

- lower temperature
- plasma damage
- higher pressure
- better step coverage





### **Step Coverage**

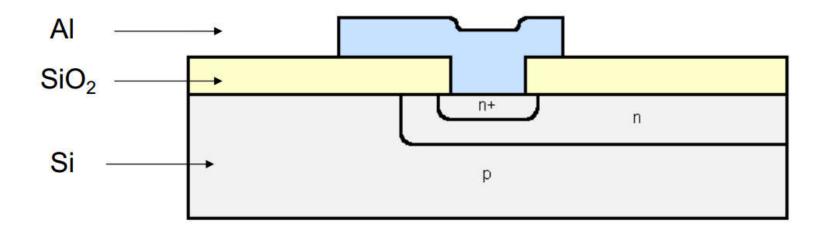


surface reaction

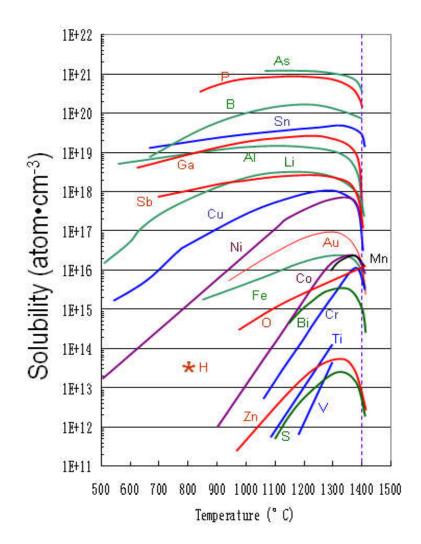
#### ballistic transport

#### Al has good adhesion on Si and SiO<sub>2</sub>

- Al has high solubility in Si
- $\Box AI + SiO_2 = AI_2O_3 + Si$

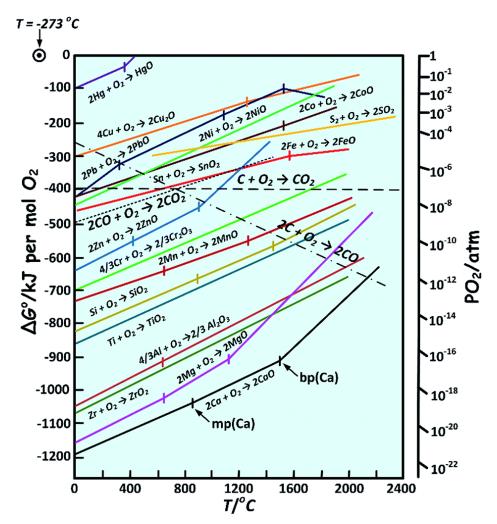


#### **Q: How about Cu and Au?**



#### solubility of metals in Si

#### Ellingham diagram

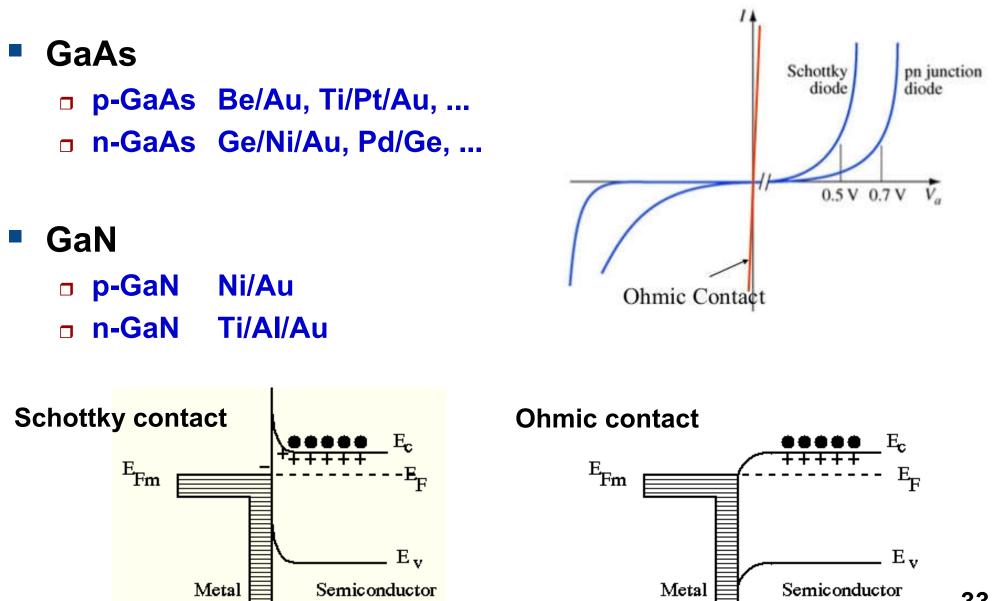


#### **Formation of metal oxides**

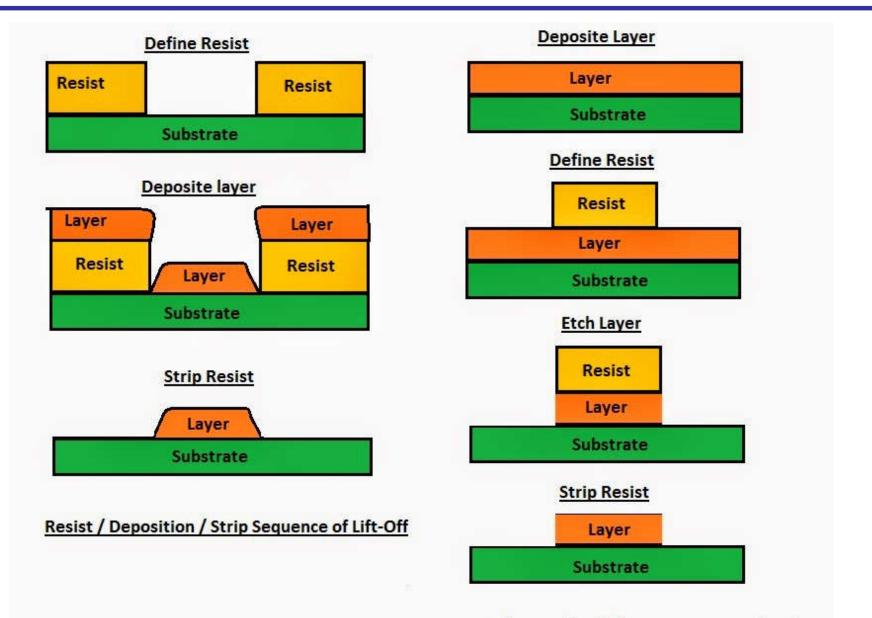
- Metals like Ag and Au tend to have poor adhesion on Si and SiO<sub>2</sub>
- Substrate clean
- Deposit a thin (~5 nm) Ti or Cr layer for adhesion
- Plasma treatment
- Monolayer bonding



## **Typical Ohmic Contacts for III-V**

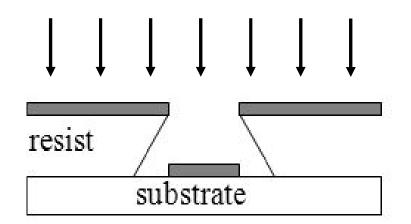


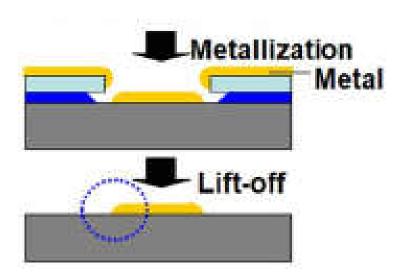
### **Thin Film Patterning**



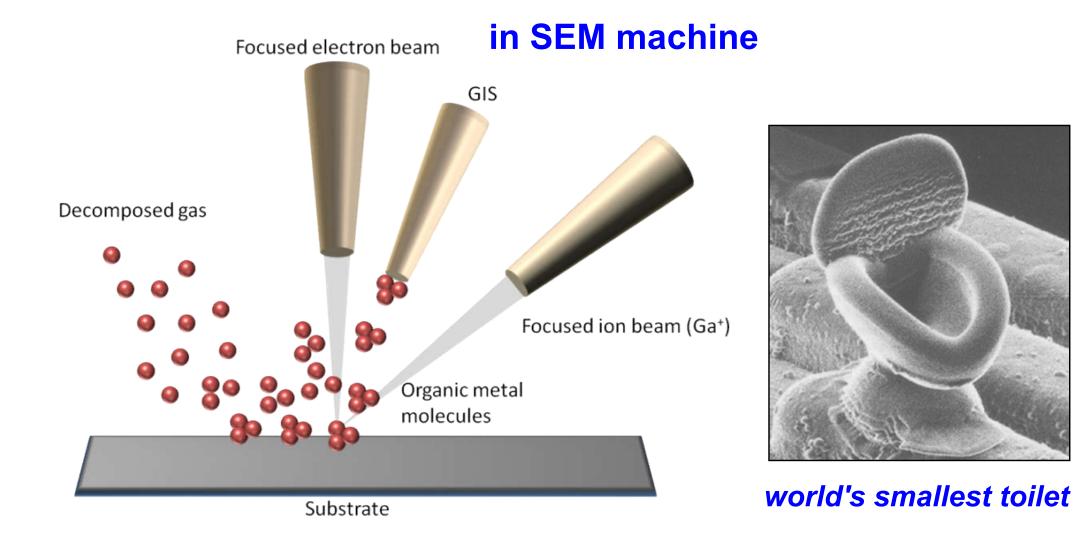
## **Requirements for Liftoff**

- Avoid Step Coverage
- PVD instead of CVD
  avoid high temperature
- Photoresist (PR) process
  - negative PR preferred (Why?)
  - increase PR thickness
  - multilayer PR





### **Focused Ion Beam (FIB) Deposition**



Etch: Ga

**Deposition: Pt**